## Towards annealing-stable molybdenum-oxide-based hole-selective contacts for silicon photovoltaics

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Molybdenum oxide (MoO<sub>x</sub>) combines a high work function with broadband optical transparency. Sandwiched between a hydrogenated intrinsic amorphous silicon passivation layer and a transparent conductive oxide, this material allows a highly efficient hole-selective front contact stack for crystalline silicon solar cells. However, hole extraction from the Si wafer and transport through this stack degrades upon annealing at 190 °C, which is needed to cure the screen-printed Ag metallization applied to typical Si solar cells. Here, we show that effusion of hydrogen from the adjacent layers is a likely cause for this degradation, highlighting the need for hydrogen-lean passivation layers when using such metal-oxide-based carrier-selective contacts. Pre-MoO<sub>x</sub>-deposition annealing of the passivating a-Si:H layer is shown to be a straightforward approach to manufacturing MoO<sub>x</sub>-based devices with high fill factors using screen-printed metallization cured at 190 °C.

Passivating carrier-selective contacts offer a simple way to approach the theoretical efficiency limit of silicon (Si) solar cells without the need for expensive layer patterning, and they can offer superior performance under outdoor conditions [1,2]. In conventional Si homo-junction solar cells, carrier separation is ensured by highly doped regions within the Si wafer, and electrical contacts with low resistivity are obtained by applying metal electrodes directly onto the highly doped Si surfaces. Despite being extensively manufactured worldwide, such solar cells are limited in their efficiency potential due to defect-assisted and Auger recombination of charge carriers, respectively at the metal/silicon interface and in the doped Si regions. The use of passivating contacts-like the ones used in silicon heterojunction (SHJ) solar cells [3]—suppresses these recombination routes by separating the metal from the surface of the Si wafer and omitting heavy doping of the Si wafer. To reduce further optical losses in passivating contacts, the application of wide-bandgap materials like molybdenum-, tungsten-, titanium- or nickel-oxide (MoO<sub>x</sub> [4-6], WO<sub>x</sub> [7, 8], TiO<sub>2</sub> [9-11], NiO [10]), as well as lithium- or magnesium-fluoride (LiF [12], MgF<sub>2</sub> [13]) has received much attention in recent years. These materials feature a work function that is either higher than the ionization energy, or lower than the electron affinity of crystalline Si (c-Si). Therefore, when in contact with Si and depending on the band lineup and defect density of the interface, these materials may induce an electrical potential at the Si surface, which promotes the collection of either only holes or only electrons. As a specific example, the transition metal oxide  $MoO_X$  (x  $\approx$ 3) combines a wide optical bandgap energy of 2.8-3.1 eV [14] with a high work-function of 4.8 to 6.9 eV [15, 16]; when deposited on Si surfaces, it may thus promote hole collection. Practically, conversion efficiencies up to 22.5% [5] were demonstrated for Si solar cells employing a MoO<sub>X</sub> (x  $\approx$ 3) based hole-selective front emitter stack. The higher transparency of MoO<sub>X</sub> compared to p-type hydrogenated amorphous silicon layers (a-Si:H) led to ~0.3 mA/cm<sup>2</sup> gain in photo current density compared to the reference SHJ cells [5]. To ensure passivation of the silicon surface, an additional a-Si: H buffer layer was inserted underneath this MoO<sub>x</sub> film, similar as in conventional SHJ technology. The MoO<sub>x</sub> film was capped with a transparent conductive oxide (TCO), either hydrogen doped indium oxide (IO:H) and indium tin oxide (ITO) stack [17] or simply ITO, to minimize resistive losses and maximize light in-coupling [18]. To finish the devices, a Cu front grid electrode [19 was formed by electroplating. This metallization technique notably does not require any thermal treatment above 125 °C. Despite its high work function, MoO<sub>x</sub> is an n-type material [20]. As a consequence, efficient carrier extraction requires that photogenerated holes in the valence band of c-Si recombine with electrons present in the MoO<sub>X</sub> conduction band; the latter electrons are injected from the degenerately n-doped TCO [4, 6]. Efficient charge-carrier transport through this contact stack depends on the thickness, defect density (for trap-assisted transport) and work function of MoO<sub>X</sub> [21, 22], as well as on the line-up with the band edge energies of the surrounding layers.

Despite these promising results, industrial implementation of such contacts demands their compatibility with contemporary high-throughput grid metallization techniques, which currently consist mainly of screen-printing Ag paste, followed by a moderate-temperature cure at about 190 °C. Unfortunately, applying this to our current implementation of  $MoO_X$ -based solar cells results in light current-voltage (JV) curves that are S-shaped near the open-circuit voltage (V<sub>OC</sub>), resulting in reduced fill factors (FF) below 70% [5]. We present here an experimental investigation by means of thermal desorption spectroscopy (TDS) and surface photovoltage spectroscopy (SPV) of the underlying degradation mechanisms leading to this FF degradation. These results hint at hydrogen effusion from the a-Si:H layer being the origin of the FF degradation, and we propose a solution similar to the approach described in [23] to mitigate this effect by annealing the a-Si:H-coated silicon wafers prior to  $MoO_x$  deposition: This reduces the hydrogen content in the film, leading to reduced H effusion upon the final annealing step of the finished device.

For all experiments, 240  $\mu$ m thick, ~3  $\Omega$ cm phosphorus-doped, float-zone (100) Si wafers were textured and cleaned. During solar cell fabrication at EPFL, a 5-nm-thick intrinsic, a-Si:H layer was applied by plasma enhanced chemical vapor deposition (PECVD) on the wafer's front side, and a ~15 nm thick stack of intrinsic and n-type doped a-Si:H layers was deposited as the rear-electron collecting contact. On the front side, a ~8 nm thick MoO<sub>X</sub> layer was thermally evaporated from MoO<sub>3</sub> powder using a deposition rate of 0.05 nm/sec. The active solar cell areas were defined by sputter deposition of ~70 nm thick ITO layers on the MoO<sub>X</sub>. A full area ITO/Ag rear contact stack was deposited by sputtering and finally a front metal grid was prepared by Ag screen printing. Current-voltage (JV) characteristics were measured after stepwise curing of the Ag paste at 130 °C and 190 °C. SPV measurements were performed at Helmholtz-Zentrum Berlin (HZB) on cells with Cu-plated front contacts using a home-built setup with 905 nm laser excitation [24]. TDS was performed at the National Institute of Advanced Industrial Science and Technology (AIST) with a constant heating rate of (20.0 ± 0.1) K/min at a base pressure lower than 10<sup>-9</sup> mbar.

Figure 1a shows the recorded TDS spectra of  $H_2$  from single a-Si:H and MoO<sub>X</sub> films as well as a-Si:H/MoO<sub>X</sub> and a-Si:H/MoO<sub>X</sub>/IO:H stacks. A single a-Si:H layer releases hydrogen (H<sub>2</sub>) already at low temperatures around 100 °C with a desorption peak at 360 °C, similarly to literature [25-28]. The presence of a MoO<sub>X</sub> layer on top of the a-Si:H leads to an earlier and also more pronounced release of H<sub>2</sub> (at temperatures as low as 150 °C) and shifts the effusion peak to ~320 °C. A similar effect was obtained from identical measurements employing doped a-Si:H overlayers [28]. This effect was explained by reduction of the defect-formation energy when the Fermi-level inside the intrinsic a-Si:H is shifted closer to its band edges. Our experiments on MoO<sub>X</sub>-based devices support well these earlier findings since a Fermi-level shift in the intrinsic a-Si:H closer to the valence band is also expected in this case, though the surface from which H effuses is different in our case. Next, we find that the H<sub>2</sub> effusion peak of the a-Si:H/MoO<sub>X</sub>/IO:H stack is

significantly lower in intensity, with an onset at increased temperatures (~230 °C vs. ~170 °C for the a-Si:H/MoO<sub>X</sub> stack). This suggests that H<sub>2</sub> from the a-Si:H/MoO<sub>X</sub> stack is partially absorbed in the IO:H, and released in the form of H<sub>2</sub>O, as clearly observed in Fig. 2b. The H<sub>2</sub>O desorption from IO:H (Figure 2b) is in good agreement with refs. [29,30]. The H<sub>2</sub>O effusion spectra of c-Si with MoO<sub>X</sub> layers and a-Si.H/MoO<sub>X</sub> stacks have a sharp rise at temperatures close to 75 °C and indicate the thermal decomposition (reduction) of MoO<sub>X</sub> (x~3), which is partly triggered by the presence of hydrogen [31]. The effect of the H<sub>2</sub> effusion on the solar cell performance will be discussed below, based on SPV and IV measurements of completed solar cells.

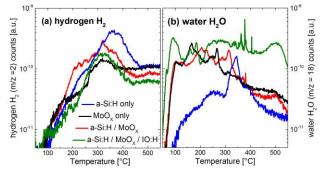


FIG. 1. H<sub>2</sub> and H<sub>2</sub>O thermal desorption spectra of test structures consisting of a single layer of MoO<sub>x</sub> or a-Si:H on c-Si wafers or of stacks of MoO<sub>x</sub>/a-Si:H or a-Si:H/MoO<sub>x</sub>/IO:H on c-Si wafers. H<sub>2</sub> effusion spectra: the existence of a MoO<sub>x</sub> layer on top of a-Si:H (red curve) leads to a shift of the main hydrogen effusion peak to lower temperatures as compared to a-Si:H alone (blue curve). H<sub>2</sub>O effusion spectra: the spectra show the release of H<sub>2</sub>O from the top IO:H film (green curve) and indicate the decomposition (reduction) of the MoO<sub>x</sub> films (red and black curves).

Figure 2a shows the changes in the MoOx[/a-Si:H]/c-Si band bending of our solar cells throughout annealing (temperatures of 100-250 °C, each step 5 min), as extracted from SPV measurements. Note that although the absolute value obtained with the setup used in this study are typically lower compared to the ones measured in other places [23]. relative differences between samples can be discussed and correlate well to device properties [8]. The band bending of the MoO<sub>x</sub>-based solar cell with the a-Si:H(i) buffer layer is significantly reduced by annealing, whereas the changes in the MoO<sub>x</sub> cell without the a-Si:H(i) layer are much smaller (especially between 170 °C and 210 °C). The similar onset temperatures for band-bending reduction and H effusion from a-Si:H / MoO<sub>X</sub> stacks suggests that these effects are linked. We surmise that, similar to what has been found in tungsten oxide  $(WO_X)/a$ -Si heterojunctions [8], hydrogen effuses from the a-Si:H layer, partially reducing MoO<sub>X</sub> and lowering its workfunction, leading to reduced c-Si band bending, thus degrading the hole-selectivity of our contact. This is reflected in the IV curves of corresponding devices at various annealing temperature, where a strong S-shape characteristic appears after annealing at 190 °C for the  $MoO_X$ based device incorporating an a-Si:H layer but not for the one not incorporating this a-Si:H layer. Also, in case the degradation of workfunction is accompanied with a lowering of the electron affinity and ionization potential, the energetic gap between the a-Si:H valence band and the TCO conduction band is increased, as sketched in Figure 2b, deteriorating transport by reducing the probability of (trap-assisted) tunneling within the MoO<sub>X</sub> layer. Notably, reduction of the oxidation state of the MoO<sub>x</sub> layer-and possibly hydrogenation-can- affect the workfunction by moving the Fermi-level only without affecting the band structure, but can also reduce the workfunction and activation energy and ionization potential similarly, thus shifting the band structure towards the vacuum level energy [15,32]. Both effects have been reproduced in Fig. 2b to keep it general. A reduced band bending close to the Si surface also decreases the potential drop in the a-Si:H layer. In turn, the thermionic field emission over the band offset at the a-Si:H/c-Si interface is reduced, eventually leading to a transport barrier for holes, resulting in S-shaped JV curves as reported in our earlier work [5]. The dominant transport limitation at stake in our device is unclear—the role of eventual trap states and dipoles possibly bringing additional contributions, though the origin being a drop of workfunction through H-enhanced reduction is a likely cause in all three cases which correlates well with effusion measurement. As a consequence, buffer layers which do not effuse hydrogen upon annealing up to 200 °C are desirable to obtain annealing-stable MoOx-based hole-selective contacts. The next section discusses how annealing the a-Si:H passivating layers prior to MoO<sub>X</sub> deposition could lead to such H-effusion-free buffer layer, as we observe good passivation and efficient carrier transport when using such layer—even after annealing the finished device at 190 °C.

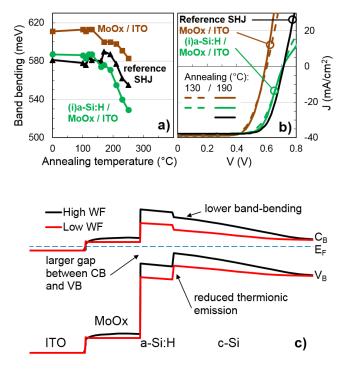


Figure 2. (a) Band bending extracted from SPV measurements on three solar cells with MoO<sub>X</sub>based hole-selective front contacts and one SHJ reference cell. The cell without the a-Si:H buffer layer has the strongest band bending and is less affected by stepwise annealing. (b) Current-voltace characteristics of corresponding devices after annealing at 130 °C and 190 °C. (c) Qualitative sketch of the energy band diagram of a MoOx-based front contact. The effect of a reduced MoO<sub>X</sub> work function on the MoOx/a-Si(i)/c-Si(inverted n-type surface layer) stack is illustrated (VB= valence band,  $E_F$  = Fermi-energy, CB= conduction band).

Fig. 3 shows the electrical properties of solar cells incorporating a pre-MoO<sub>x</sub>-deposition annealing step at a temperature of 200 °C to 300 °C. The parameters are displayed after two different temperatures of annealing of the finished device, used to cure the screen printed Ag contacts: 130 °C and 190 °C. The Voc is degraded by the pre-MoOx-deposition annealing, with a stronger drop when increasing the pre-annealing temperature. This can be attributed to dehydrogenation and thus passivation degradation of the a-Si:H passivation layer, both on the holecollecting side and electron-collecting-side for highest temperatures [28,33]. Notably, a degradation is seen after metal curing at 190 °C for devices exposed to a pre-MoOx-deposition annealing at a temperature below 250 °C, whereas an improvement is observed for temperatures higher than 250 °C. This can possibly be attributed to the recovery of sputter-induced damage, occurring for all samples but specifically visible for the higher pre-MoOx-deposition annealing temperatures: for the low pre-MoOx-deposition annealing temperatures, this recovery is overcompensated by a drop due to the loss of selectivity of the MoOx-based device. Turning to FF, an optimum pre-annealing temperature of 250 °C is seen both prior to and after final curing at 190 °C. The drop for too high pre-MoOx-deposition annealing temperatures can be attributed to the passivation loss due to too strong de-hydrogenation, whereas the drop for too low temperatures can be attributed to the selectivity drop, specially seen after curing at 190 °C. A 4% FF gain is observed upon pre-annealing at 250 °C compared to pre-annealing at 200 °C only (which gives similar results to no pre-annealing, not shown), clearly evidencing the effectiveness of the approach. No trend is seen for the J<sub>SC</sub> within the accuracy of the measurement, and finally the efficiency follows mostly the FF trend with an optimum at a pre-annealing temperature of 250 °C. Using this pre-annealing temperature, we fabricated solar cells with MoO<sub>X</sub>-based hole-selective contacts reaching fill factor values of 76% and an efficiency of 20.8% after curing of the front Ag metal grid at 190  $^{\circ}$ C. Notably, the corresponding JV-curve does not exhibit an S-shape around V<sub>OC</sub> and the determined FF is close to the value achieved with sister samples featuring our baseline a-Si:H(i)/a-Si:H(p) front hole-selective contact. The slightly reduced  $V_{OC}$  compared to values obtained with standard a-Si:H(p) layers can be attributed to a slightly degraded a-Si:H passivation from pre-annealing, which may be resolved by designing passivation layers more resilient to thermal annealing at 250 °C, or releasing less hydrogen at this temperature, such as a-SiC<sub>X</sub>:H films [33].

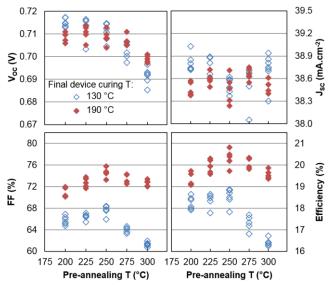


FIG. 3. JV parameters of MoOx-based solar cells as a function of the temperature of pre-MoOxdeposition annealing of the a-Si:H layers. JV parameters are displayed after curing the finished device at 130 °C and at 190 °C.

To conclude, our experiments show that silicon solar cells with a-Si:H/MoO<sub>X</sub>/TCO hole-selective contact stacks suffer from severe fill factor degradation and S-shaped JV curves when thermally annealed at 190 °C. Though the presented results are not sufficient to fully explain this degradation, we show in this study that it can be mitigated by reducing the amount of hydrogen in the adjacent layers. TDS measurements provide evidence that a MoO<sub>X</sub> overlayer shifts hydrogen effusion from a-Si:H(i) layers towards lower temperatures, confirming the theory of Fermi-level-dependent hydrogen bond breaking. SPV measurements showed that insertion of the a-Si:H(i) passivation layer and annealing at 190°C leads to a lower band bending and indicate that the work function of MoO<sub>X</sub> decreases with increasing annealing temperature. In summary, our observations hint at a transport barrier induced by hydrogen release from the a-Si:H layer, and we suggest a pre-MoO<sub>X</sub>-deposition annealing step of the a-Si:H layer to reduce its hydrogen up to 20.8%-efficient MoO<sub>X</sub>-based solar cells, using a Ag-paste curing temperature of 190 °C. Yet, although such thermal treatment allows significant improvement compared to a non-annealed device, the efficiency of MoO<sub>X</sub>-based devices obtained with this approach is still limited by a passivation/transport trade-off, highlighting the need for an alternate passivation strategy to fully exploit the potential of MoO<sub>X</sub> as a hole-selective contact.

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